Metal concentrations in razor clam *Solen dactylus* (Von Cosel, 1989) (Bivalvia: Solenidae), sediments and water in Golshahr coast of Bandar Abbas, Persian Gulf

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Abstract
In a complementary field study, the concentrations of some metals (Cu, Ag, Pb, Zn, Ni, Co, Mn, Fe, As, Cd, Cr, Mg and Ba) were measured in clam *Solen dactylus*, sediments and water collected at two stations (Park-e-Qadir, 56°20’E, 27°11’ and Nakhl-e-Nakhoda, 56°23’E, 27°10’N) of Bandar Abbas coastal waters of the Persian Gulf in November 2008 and February 2009 showing different contamination levels. Although there is little information on metal concentrations in macro-benthic animals in this area, this study, for the first time, reports the accumulation of some metals in *S. dactylus* in order to introduce this species as a bioindicator for this area. Results indicated that Mg and Fe were the two most abundant metals in clams and sediments. The maximum and minimum metal concentrations in clams belonged to Mg (3850-5040 µgg⁻¹ dry wt) and Ag (0.30-0.40-0.58 µgg⁻¹ dry wt), respectively. There was a significant relationship between the accumulation of metals in clams, sediment and water samples. A significant relationship between clam lengths and concentrations of Cu (positive) and Mg (negative) were observed. Our study also showed that variable metal concentrations were related to different sampling stations, seasons and their interactions as well. Bioaccumulation of metals in clams was significantly different for eight metal elements between start of the gametogenesis and ripeness stages. Our investigation indicated that the clam *S. dactylus* could be a useful bioindicator for Zinc.

Keywords: Metal, Bioaccumulation, Bioindicator, *Solen dactylus*, Sediment, Water, Persian Gulf

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Introduction

The Persian Gulf is one of the largest oil and gas resources in the world. This area represents a tropical case study mainly due to intensive oil extraction and related activities which can affect its pollution level. The industrial and urban activities in the tropical coastal zones increase the release of contaminants, which can be a threat for the marine ecosystems and have effects on marine diversity locally and globally (Metian et al., 2008).

In our studies, we are focusing on the determination of metal concentrations in some coastal marine species, particularly, in Bivalves and use of these organisms as bioindicators. This kind of approach has been extensively applied in other marine environments using molluscs, oysters, as well as bivalve species. For example, Etim et al. (1991) have reported temporal trends in some heavy metal concentrations such as Zn, Ni, Pd, and Cd in the Cross River, Nigeria using the clam Egeria radiata. They showed that concentration of Ni and Zn was lower than the maximum permissible levels concentration in this area. Bilos et al. (1998) determined some metal concentrations including Cu, Cr, Mn, and Zn in Asiatic clams (Carbicula fluminea) which were collected at the Río de la Plata coast, Argentina to assess the amount of metal pollution in the area and compared to those reported for the other moderately polluted world rivers. Lu et al. (2005) worked on bioavailability of metals in sediment from northern San Francisco Bay and survival of clam Macoma nasuta. They concluded that survival of clams in sediments with high concentration of metals was decreased. Pourang et al. (2005) studied the trace element concentrations including Cd, Pb, Ni and V in fish, superficial sediments and water from northern parts of the Persian Gulf. They found that Cd concentration in sediment of all Northern parts of the Persian Gulf was higher than Regional Organization for the Protection of the Marine Environment (ROPME) Sea Area Standard. Metian et al. (2008) worked on accumulation of metals and metalloid in scallop Comiptopallium radula in New Caledonia. They demonstrated that Zn was mainly concentrated in the digestive gland (65%) and Co in Kidneys (81%). They concluded from their research that C. radula could be a valuable local biomonitor species for Ag, As, Cd and Fe.

Yap et al. (2009) studied the concentration of some heavy metals in snail Telescopium telescopium as a biomonitor in the tropical intertidal area in Peninsular, Malaysia. In their study, digestive gland had higher Zn concentration and shells demonstrated higher Pb concentration. Chouvelon et al. (2009) reported Hg concentrations and their risk assessment in several marine organisms in New Caledonia. Their study showed that Hg concentration in this area is high which would be dangerous to sea food consumers. Recent studies by Hédouin et al. (2006 and 2009) screened metal concentrations in a variety of bivalves, mainly oysters Isognomon isognomon and the edible clam Garfarium tudium showed that those marine organisms satisfied the...
We are particularly interested in the determination of metal concentrations in some marine organisms such as sessile benthic molluscs which inhabit in the Iranian coastal waters of the Persian Gulf. These organisms can be exploited for contamination determination of the area as potential bioindicators. The advantage of using such organisms for biomonitoring plans has been reported (Yap et al., 2009). *Solen dactylus* is an edible sessile clam which inhabits intertidal sandy-muddy beaches along the Oman Sea and the Persian Gulf (Saeedi et al., 2009). Heavy metal accumulations in edible marine organisms have always been threats for their consumers. In order to assess the magnitude of trace metal pollution in the Northern parts of the Persian Gulf, we have begun a research program aimed at investigation and recognition of coastal marine organisms as bioindicator species. The aim of this study was, therefore, to provide baselines information on the metal ion contamination status of the Persian Gulf coastal marine environment. A wide range of metal element bioaccumulations were determined in *S. dactylus*, water and sediments, and also the relationships between lengths and sexes of clams and the metal concentrations in them were studied. This study also determines the differences between metal concentrations in clam at two stations of the Iranian coastal waters of the study area in two different seasons (November 2008 and February 2009).

### Materials and methods

#### Study area and sampling

Two sampling stations along the Golshahr coast of Bandar Abbas in the Persian Gulf were selected according to the contamination status by direct observation. Pollutants (e.g. sewage and urban wastes) in the first station (Park-e-Qadir, $56^\circ$20′E, $27^\circ$11′N) were more than the second station (Nakhl-e-Nakhoda, $56^\circ$23′E, $27^\circ$10′N) (Figure 1). Anthropogenic inputs and occurrence of rubbish dump make the first station as a polluted station. Specimens of *S. dactylus* were collected by hand with a 0.5 m long metal wire forming a V shape at one end (Saeedi et al., 2009) during low tide. The sampling was done in November 2008 (start of gametogenesis, stages I and II) and February 2009 (ripeness stage, stage III A) (Saeedi et al., 2009) along two intertidal stations of Golshahr coast in Bandar Abbas. The specimens were kept in seawater of the sampling station for 24 hrs in order to purify or depurate the gut contents and mantle cavity (Hédouin et al., 2009), then they were frozen at -20°C in a refrigerator in Shahid Beheshti University Biosystematics’ lab for future analysis. 150 to 300 g sediments (top 5 centimeter layer) were collected by acid-washed containers in the sampling stations and immediately transferred to the laboratory in acid-washed plastic bags and kept at -20°C for future metal analyses (Chen et al., 2007). 300 to 600 ml seawater samples were collected from the sea-surface at <1m depth by acid-washed containers and kept in the laboratory at 4°C.
Figure 1: Field study area in Bandar Abbas, Persian Gulf. Inset shows the sampling sites: ‘Park-e-Qadir’ and ‘Nakhl-e-Nakhoda’

Biometry and sample preparation for metal analysis
In total, 240 clams (n=60 per station per sampling month) were collected and studied during this study. The biometrical measurements including anterio-posterior length (length), dorso-ventral length (width) and the distance between two valves (diameter) to the nearest 0.1 mm and total weight (TW), wet weight of the soft parts (SPW) to the nearest 0.1 mg were measured using vernier calipers and a digital balance, respectively (Saeedi et al., 2009). Only adult samples with shell lengths longer than 45 mm were selected for analysis. Soft parts of clams were removed from the shells. Six length classes from 48 to 108 mm were selected and 10 clams for each one were pooled, weighed (wet wt) (Maanan, 2008), and dried at 60°C in the oven until reaching the constant weight (Hédouin et al., 2009). In February 2009, females, males and sex undetermined were pooled separately because we wanted to clarify metal concentration in different sexes. Dried samples weighed again (dry wt) and ground in an agate mortar for homogenization (Maanan, 2008). Sediments were dried at 60°C until reaching the constant weight, and then were sieved by a 1 mm mesh size before metal analysis (Hédouin et al., 2009). Dried biological samples of clams from 0.1 to 0.5 g and dried sediment samples about 0.5 g were digested using a 3:1 (v/v) 65% HNO₃ 30% HCL mixture at room temperature overnight, then using a microwave for 30 min with constantly increasing temperature up to 150 °C for biological samples and up to 100 °C for sediment samples, then 15 min at this
maximum temperature. All samples were then diluted to 30-50 ml with milli-Q water (Hédouin et al., 2009).

About 100 ml of water samples were filtered through 0.4 µm filter papers. Finally, elements were analyzed using an atomic absorption spectrophotometer certified ISO/IEC 17025. The sample concentrations are presented as µgg⁻¹ dry wt (Yap et al., 2009) for clams and sediments and µgl⁻¹ for water samples. For verification of the quality of metal analysis, all samples were reanalyzed in Acme Labs in Vancouver, Canada and used SRM (Standard Reference Materials).

Environmental factors measurement

Sea-surface temperature, Salinity, dissolved oxygen and pH were measured with 3 replicates at low tide during the study period separately in both stations.

Statistical analysis

SPSS version 17 and Excel 2007 software were used for statistical analysis and plotting all graphs. All data were checked for normality (Shpiro-wilk test) to use in future parametric tests. A two-way analysis of variance (ANOVA) was used to determine the effects of sampling season (sexual and rest stages) and to study stations as fixed factors on the variation of metal concentrations in the soft parts (Maanan, 2008). Pearson correlation between size and total different metal concentrations in soft tissues in each length class were used. Paired samples t-test was used to determine the differences of metal concentrations in clams in proliferation of gonad and ripeness stages (Chouvelon et al., 2009). An independent sample t-test was used to determine the significant differences of metal concentrations between male and female clams as well as metal concentrations in sediment and water between two stations. A non-parametric test Spearman’s correlation was used to determine the relationship between the metal concentration in clams, sediment and water. The level of significance was set at α=0.05.

Result

Metal concentrations in Clam Solen dactylus, sediment and water

A total of 240 clams, were collected and studied during this work (52 to 108 mm in length, 2.56 to 21.15g in weight). The element concentrations in S. dactylus, sediment and water are collected in tables 1 and 2. Concentrations of some metal elements in clams were under the detection limit (indicated as < in all tables) which restricted comparisons between them in the two stations. The maximum and the minimum average (±SD) of metal concentrations in clams in station 1 belonged to Mg (3850±589.06 in November 2008 and 5040±811 µgg⁻¹ dry wt in February 2009) and Ag (0.58±0.23 in November 2008 and 0.40±0.00 µgg⁻¹ dry wt in February 2009), respectively. As it is also shown in table 1, the maximum and the minimum averages of metal concentrations in sediment and water samples belonged to Mg and Ag, respectively. In general, the most abundant and typical metals in clams and
Sediments were Mg and Fe and in water it was Mg and Cu. The Zn concentrations in the clams were more than that in sediment and water samples, while for Pb and Fe the concentrations in sediments were higher than those in clams and in water (Tables 1 and 2). Also the average concentrations of Cu were high first in water (101±6.18 µg l⁻¹) and then in clam (8.50±0.70 µg g⁻¹ dry wt) and finally in sediment (6.75±1.50 µg g⁻¹ dry wt). Therefore, it is concluded that this clam might be a suitable bioindicator for Zinc based on the metal concentrations in clams more than their environment (Tables 1 and 2). In both the first and the second stations, there was a significant positive correlation between the metal concentrations in clam and sediment (Spearman’s correlation, r=0.56 and 0.67, p≤0.05), clam and water (r=0.61 and 0.86, p≤0.05) and sediment and water (r=0.66 and 0.68, p≤0.05). Figure 2 shows the mean of metal concentrations in clams at two different stations. The mean concentration of Mg in clams at Park-e-Qadir (station 1) was somehow similar (4594±929.58 µg g⁻¹ dry wt) to Nakhl-e-Nakhoda (station 2) (4585±826.48 µg g⁻¹ dry wt). The mean concentrations of Fe and Mn at station 2 in clams (1662±711.26 and 48±22.15 µg g⁻¹ dry wt) were higher than those at station 1 (494±428 and 16±12.53 µg g⁻¹ dry wt), whereas the mean concentration of Zn at the first station (81±16.38 µg g⁻¹ dry wt) was greater than that at the second station (61±6.43 µg g⁻¹ dry wt). Other metal concentrations in clams show little differences between two stations. Figure 3 presents the metal concentrations in sediments at both stations 1 and 2. The mean concentration of Mg in sediments at station 1 (15001±425 µg g⁻¹ dry wt) and station 2 (14851±6498 µg g⁻¹ dry wt) showed a little difference. The mean concentrations of Fe and Mn at the second station (13801±5190 and 504±215 µg g⁻¹ dry wt, respectively) were considerably higher than those at the first station (9051±1203 and 481±12 µg g⁻¹ dry wt, respectively). As it is shown in Figure 4, the mean concentration of Mg in water at two stations was nearly equal. The relationship between all metal concentrations in soft tissue of clams were studied and it was found that only Fe and Mn concentrations showed a significant positive correlation (r²=0.97, p≤0.05) (Figure 5).

Figure 2: Average metal concentrations in clam at two different stations; Station 1, Park-e-Qadir and Station 2, Nakhl-e-Nakhoda, Bandar Abbas (Standard Bar: SE)
Figure 3: Average metal concentrations in sediment at two different stations; Station 1, Park-e-Qadir and Station 2, Nakhl-e-Nakhoda, Bandar Abbas (Standard Bar: SE)

Figure 4: Average metal concentrations in water at two stations; Station 1, Park-e-Qadir and Station 2, Nakhl-e-Nakhoda, Bandar Abbas (Standard Bar: SE)

Figure 5: Linear regression relationship between two metal elements; Mn and Fe, in S. dactylus, Bandar Abbas
Table 1: Element concentrations in different length classes of *Solen dactylus* (mean±SD; µgg⁻¹ dry wt, n=10), sediment (mean±SD; µgg⁻¹ dry wt, n=3) and water

| Lengt Class | Mean Length (mm) | Weigh t (g) | Cu | Ag | Pb | Zn | Ni | Co | Mn | Fe | As | Cd | Cr | Mg | Ba |
|-------------|-----------------|------------|----|----|----|----|----|----|----|----|----|----|----|----|----|----|
| Park-e-Qudir | 1 64.8±2 | 5.33±0.8 | 24.00±2.0 | 0.30±0.0 | <3 | 63.00±10.9 | 6.00±0.8 | 5.00±0.4 | 39.00±0.4 | 110.00±135 | 25.00±4.0 | <0.50 | 5.00±0.4 | 4600.00±930 | 0.0 | 8.00±1.0 |
| 2 70.83 ±0.6 | 6.59±1.7 | 12.00±1.7 | 0.50±0.0 | <3 | 60.00±12.0 | 4.00±0.0 | 2.00±0.0 | 24.00±2.0 | 900.00±73.2 | 26.00±5.0 | <0.50 | 4.00±0.0 | 3000.00±73.2 | 0.0 | 7.00±0.0 |
| 3 74.16 ±1.3 | 7.23±3.8 | 23.00±2.0 | <0.3±0.0 | 5.50±0.9 | 6.00±0.0 | 4.00±0.4 | 39.00±7.0 | 150.00±127.0 | 30.00±5.0 | <0.50 | 4.00±0.0 | 3800.00±134 | 0.0 | 7.00±0.0 |
| 4 80.91 ±1.7 | 9.30±2.0 | 21.00±3.8 | <0.3±0.0 | 63.00±11.0 | 5.00±0.0 | 4.00±0.0 | 34.00±3.0 | 1000.00±104.0 | 25.00±2.0 | <0.50 | 4.00±0.3 | 3800.00±692.0 | 8.00±0.0 |
| 5 91.91 ±2.4 | 12.37±2.6 | 18.00±1.7 | <0.3±0.0 | 65.00±13.0 | 4.00±0.0 | 2.00±0.0 | 27.00±6.0 | 800.00±8.0 | 30.00±3.0 | <0.50 | 3.00±0.3 | 3500.00±56.0 | 6.00±0.0 |
| 6 106.1 ±1.7 | 19.57±4.6 | 23.00±2.1 | <0.3±0.0 | 73.00±11.0 | 3.00±0.0 | 1.00±0.0 | 16.00±2.0 | 500.00±9.0 | 26.00±6.0 | <0.50 | 3.00±0.5 | 3100.00±592.0 | 21.00±7.0 |
| Sediment | ---- | ---- | 5.00±0.0 | <0.3 ±0.0 | 4.00±0.0 | 14.00±0.0 | 29.00±7.0 | 13.00±1.0 | 472.00±29.0 | 8200.00±376.0 | 10.00±2.0 | <0.50 | 33.00±9.0 | 14700.00±2963.0 | 54.00±12.0 |
| Water | ---- | ---- | 9.46±2.1 | 0.14±0.0 | 1.60±0.0 | 13.20±0.0 | 5.60±0.0 | 1.87±0.0 | 45.99±1.9 | 25.00±7.9 | 83.70±19.0 | 1.00±0.0 | 5.60±1.2 | 1277689.0±2981.0 | 15.18±5.4 |

| Nakhle-e-Nakhoda | 1 60.66 ±4.5 | 4.00±0.0 | 12.00±2.4 | 0.70±0.0 | 27.0±0.0 | 76.00±19.0 | 9.00±8.0 | 25.00±1.0 | 50.00±1.0 | 1900.00±293.0 | 29.00±5.0 | <0.50 | 8.00±0.7 | 5500.00±873.0 | 40.00±13.0 |
| 2 66.83 ±2.0 | 5.28±0.0 | 22.00±5.0 | 0.50±0.0 | 4.00±0.0 | 62.00±17.0 | 5.00±0.0 | 3.00±0.0 | 25.00±1.0 | 900.00±9.0 | 27.00±5.0 | <0.50 | 3.00±0.1 | 3600.00±73.0 | 7.00±2.0 |
| 3 81.08 ±1.5 | 9.27±2.1 | 12.00±2.4 | <0.3±0.0 | <5.0 | 58.00±15.0 | 5.00±0.0 | 3.00±0.0 | 35.00±1.0 | 130.00±123.0 | 35.00±8.0 | <0.50 | 4.00±0.2 | 4200.00±728.0 | 8.00±2.0 |
| 4 86.16 ±1.5 | 11.83±3.6 | 10.00±1.7 | <0.3±0.0 | <5.0 | 54.00±16.0 | 6.00±0.0 | 3.00±0.0 | 45.00±1.6 | 150.00±259.0 | 27.00±7.0 | <0.50 | 6.00±0.3 | 4600.00±862.0 | 7.00±1.8 |
| 5 89.00 ±1.0 | 11.83±3.6 | 10.00±2.4 | <0.3±0.0 | <5.0 | 56.00±13.0 | 8.00±0.0 | 3.00±0.0 | 43.00±5.0 | 1600.00±162.0 | 32.00±9.0 | <0.50 | 6.00±0.3 | 42000.00±829.0 | 10.00±3.0 |
| 6 95.50 ±1.0 | 14.22±4.2 | 11.00±2.4 | <0.3±0.0 | <5.0 | 55.00±12.0 | 5.00±0.0 | 3.00±0.0 | 31.00±5.0 | 120.00±121.0 | 31.00±11.0 | <0.50 | 3.00±0.0 | 38000.00±612.0 | 6.00±0.9 |
| Sediment | ---- | ---- | 8.00±0.0 | <0.3±0.0 | 7.00±0.0 | 22.00±8.0 | 46.00±7.0 | 15.00±3.0 | 509.00±10.7 | 13600.00±2374.0 | 7.00±0.8 | <0.50 | 49.00±10.0 | 15300.00±3726.0 | 107.00±29.0 |
| Water | ---- | ---- | 102.90±26.0 | 0.20±0.0 | 0.60±0.0 | 21.70±11.0 | 10.70±2.0 | 0.46±0.0 | 49.06±1.7 | 82.00±27.0 | 78.00±31.0 | 0.44±0.0 | 10.70±2.1 | 1321297.0±3017.0 | 24.34±14.0 |

a: Concentrations <detection limit;
### Table 2: Element concentrations in different sex and length classes of *Solen dactylus* (µgg⁻¹ dry wt, *n*=10), sediment (µgg⁻¹ dry wt, *n*=3) and water (µg¹, *n*=3)

<table>
<thead>
<tr>
<th>Length Class</th>
<th>Mean Length (mm)</th>
<th>Weight (g)</th>
<th>Cu</th>
<th>Ag</th>
<th>Pb</th>
<th>Zn</th>
<th>Ni</th>
<th>Co</th>
<th>Mn</th>
<th>Fe</th>
<th>As</th>
<th>Cd</th>
<th>Cr</th>
<th>Mg</th>
<th>Ba</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Park-e-Qadir</strong></td>
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<td></td>
</tr>
<tr>
<td>1 (M)</td>
<td>86.50±1.8</td>
<td>11.49±3.6</td>
<td>15.00±3.0</td>
<td>0.40±0.0</td>
<td>5.00±0.2</td>
<td>88.00±1.7</td>
<td>3.00±0.2</td>
<td>12.00±2.0</td>
<td>9.00±0.8</td>
<td>300.00±85</td>
<td>22.00±8.7</td>
<td>0.90±0.0</td>
<td>2.00±0.0</td>
<td>3800.00±991.0</td>
<td>1.00±0.0</td>
</tr>
<tr>
<td>1 (F)</td>
<td>93.50±2.1</td>
<td>13.91±9.9</td>
<td>32.00±8.0</td>
<td>0.50±0.0</td>
<td>6.00±0.3</td>
<td>107.00±29.0</td>
<td>12.00±3.0</td>
<td>3.00±0.2</td>
<td>100.00±41</td>
<td>17.00±3.1</td>
<td>1.10±0.0</td>
<td>&lt;1*</td>
<td>5500.00±102.4</td>
<td>1.0</td>
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<tr>
<td>2 (M)</td>
<td>106.25±1.8</td>
<td>17.57±3.9</td>
<td>25.00±6.0</td>
<td>0.40±0.0</td>
<td>5.00±0.2</td>
<td>75.00±1.9</td>
<td>2.00±0.1</td>
<td>12.00±2.6</td>
<td>9.00±0.9</td>
<td>200.00±61</td>
<td>24.00±8.7</td>
<td>1.50±0.0</td>
<td>&lt;1*</td>
<td>3800.00±891.0</td>
<td>1.00±0.0</td>
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<tr>
<td>2 (F)</td>
<td>81.83±1.0</td>
<td>9.88±4.8</td>
<td>19.00±5.0</td>
<td>&lt;0.30</td>
<td>5.00±0.3</td>
<td>97.00±2.4</td>
<td>2.00±0.1</td>
<td>11.00±2.0</td>
<td>6.00±0.8</td>
<td>200.00±88</td>
<td>20.00±7.7</td>
<td>1.40±0.0</td>
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<td>5900.00±100.0</td>
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<tr>
<td>3 (M)</td>
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<td>4500.00±749.0</td>
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<td>3 (SU)</td>
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<td>&lt;0.30</td>
<td>&lt;0.30</td>
<td>88.00±1.9</td>
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<td>10.00±1.0</td>
<td>8.00±0.4</td>
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<td><strong>Sediment</strong></td>
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<td>---</td>
<td>---</td>
<td>12.00±2.0</td>
<td>&lt;0.30</td>
<td>97.00±2.7</td>
<td>2.00±0.0</td>
<td>18.00±3.0</td>
<td>8.00±0.7</td>
<td>200.00±9</td>
<td>19.00±6.0</td>
<td>1.10±0.0</td>
<td>&lt;1*</td>
<td>5800.00±1.9</td>
<td>24.0</td>
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<tr>
<td><strong>Water</strong></td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>12.00±2.0</td>
<td>&lt;0.30</td>
<td>97.00±2.7</td>
<td>2.00±0.0</td>
<td>18.00±3.0</td>
<td>8.00±0.7</td>
<td>200.00±9</td>
<td>19.00±6.0</td>
<td>1.10±0.0</td>
<td>&lt;1*</td>
<td>5800.00±1.9</td>
<td>24.0</td>
</tr>
</tbody>
</table>

### Nakhl-e-Nakhoda

<table>
<thead>
<tr>
<th>Length Class</th>
<th>Mean Length (mm)</th>
<th>Weight (g)</th>
<th>Cu</th>
<th>Ag</th>
<th>Pb</th>
<th>Zn</th>
<th>Ni</th>
<th>Co</th>
<th>Mn</th>
<th>Fe</th>
<th>As</th>
<th>Cd</th>
<th>Cr</th>
<th>Mg</th>
<th>Ba</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 (M)</td>
<td>96.50±2.1</td>
<td>92.50±1.4</td>
<td>15.00±3.0</td>
<td>0.30±0.0</td>
<td>6.00±0.3</td>
<td>57.00±1.2</td>
<td>5.00±0.3</td>
<td>9.00±0.8</td>
<td>41.00±12.0</td>
<td>1400.00±390</td>
<td>28.00±7.0</td>
<td>3.20±0.5</td>
<td>3.00±0.7</td>
<td>4300.00±721.7</td>
<td>5.00±0.2</td>
</tr>
<tr>
<td>1 (F)</td>
<td>92.50±2.1</td>
<td>11.25±4.5</td>
<td>15.00±4.0</td>
<td>0.30±0.0</td>
<td>6.00±0.5</td>
<td>66.00±16.0</td>
<td>16.00±3.0</td>
<td>25.00±7.0</td>
<td>112.00±39.0</td>
<td>3800.00±395</td>
<td>27.00±7.0</td>
<td>2.40±0.2</td>
<td>12.00±2.0</td>
<td>6800.00±1672.2</td>
<td>15.00±2.5</td>
</tr>
<tr>
<td>2 (M)</td>
<td>87.33±2.0</td>
<td>10.26±4.2</td>
<td>10.00±2.0</td>
<td>&lt;0.30</td>
<td>4.00±0.2</td>
<td>64.00±13.0</td>
<td>16.00±7.0</td>
<td>7.00±19.0</td>
<td>2100.00±725</td>
<td>26.00±6.0</td>
<td>3.20±0.4</td>
<td>7.00±0.6</td>
<td>5000.00±1000.0</td>
<td>12.00±1.0</td>
<td></td>
</tr>
<tr>
<td>2 (F)</td>
<td>81.66±2.8</td>
<td>8.84±4.0</td>
<td>12.00±3.0</td>
<td>&lt;0.30</td>
<td>5.00±0.3</td>
<td>69.00±17.0</td>
<td>6.00±3.0</td>
<td>14.00±3.0</td>
<td>45.00±18.0</td>
<td>1500.00±291</td>
<td>28.00±8.0</td>
<td>3.10±0.5</td>
<td>5.00±0.9</td>
<td>4500.00±1022.0</td>
<td>7.00±0.9</td>
</tr>
<tr>
<td>3 (M)</td>
<td>70.00±3.5</td>
<td>5.33±4.0</td>
<td>10.00±1.0</td>
<td>&lt;0.30</td>
<td>5.00±0.5</td>
<td>65.00±12.0</td>
<td>5.00±3.0</td>
<td>15.00±3.0</td>
<td>36.00±17.0</td>
<td>1300.00±294</td>
<td>27.00±7.0</td>
<td>2.90±0.4</td>
<td>4.00±0.6</td>
<td>4200.00±3982.0</td>
<td>6.00±0.7</td>
</tr>
<tr>
<td>3 (SU)</td>
<td>63.00±1.3</td>
<td>4.40±0.7</td>
<td>12.00±2.8</td>
<td>&lt;0.30</td>
<td>6.00±0.5</td>
<td>61.00±11.0</td>
<td>8.00±0.6</td>
<td>16.00±4.0</td>
<td>46.00±150</td>
<td>1700.00±429</td>
<td>26.00±10.0</td>
<td>3.20±0.5</td>
<td>5.00±0.1</td>
<td>4700.00±823.0</td>
<td>10.00±1.0</td>
</tr>
<tr>
<td><strong>Sediment</strong></td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>&lt;0.30</td>
<td>6.00±0.3</td>
<td>56.00±18.0</td>
<td>6.00±3.0</td>
<td>18.00±6.0</td>
<td>39.00±17.0</td>
<td>14000.00±263</td>
<td>20.00±9.0</td>
<td>3.20±0.3</td>
<td>4.00±0.6</td>
<td>4200.00±1066.0</td>
<td>6.00±0.3</td>
</tr>
<tr>
<td><strong>Water</strong></td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>&lt;0.30</td>
<td>7.00±0.8</td>
<td>21.00±4.0</td>
<td>45.00±14.0</td>
<td>19.00±5.0</td>
<td>498.00±27.0</td>
<td>14000.00±2893</td>
<td>7.00±0.6</td>
<td>&lt;0.5*</td>
<td>54.00±19.0</td>
<td>13200.0±3522.0</td>
<td>16.32±8.9</td>
</tr>
</tbody>
</table>

M: Male; F: Female; SU: Sex Undetermined; a: Concentrations <detection limit
Size and sex influences of clams on metal concentrations

A significant positive correlation occurred between the concentrations of Mg and Cu in clams and their lengths (r=0.63 and 0.59, \( p \leq 0.05 \)).

Metal concentrations in both male and female clams were not significantly different during the study (Independent sample t-test, df=8, \( p \geq 0.05 \)).

Metal concentrations were determined at the start of gametogenesis and the ripeness stages. A paired sample t-test was used to determine the relationship between two different sexual stages (after and before spawning). The test showed that there were significant differences between stages I and II (start of gametogenesis) and stage III A (ripeness) of gametogenesis (t=-6.18 to 4.43, df=11, \( p \leq 0.05 \)).

Differences among sampling sites and seasons

The mean concentrations of all measured elements in clams varied considerably in terms of two sampling stations, two seasons and their interactions (Two-Way ANOVA, \( p \leq 0.05 \)). Significant differences could be observed in Cu, Fe and Ba in sediment of the studied stations (Independent sample t-test, \( p \leq 0.05 \)), while the concentration of these metal elements were higher in sediments of station 2. Our analysis in water also showed that only Fe had significant differences between the two stations (t=4.76, \( p \leq 0.05 \)). Significant interaction between two seasons and two sampling stations was detected (Two-Way ANOVA, df=2, \( p \leq 0.05 \)).

Environmental factors measurement

Table 3 displays the environmental factors measurements during the periods of investigation. Temperature in February is less than November whereas other factors show a little change between two months. Therefore, due to the lack of differences in the values of environmental factors except temperature, investigations of their effects on the metals concentrations were not further pursued.

Table 3: Physical and chemical Sea-surface measurements during November 2008 and February 2009 in coastal waters of Bandar Abbas

<table>
<thead>
<tr>
<th>Month</th>
<th>Temperature (°C)</th>
<th>Salinity (psu)</th>
<th>Dissolved Oxygen (mg/l)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>November 2008</td>
<td>27</td>
<td>36.6</td>
<td>5.9</td>
<td>7.8</td>
</tr>
<tr>
<td>February 2009</td>
<td>23</td>
<td>37.2</td>
<td>5.8</td>
<td>7.6</td>
</tr>
</tbody>
</table>
Table 4: Metal concentrations (µg g⁻¹ dry wt) in some bivalves and their comparisons with the present study.

<table>
<thead>
<tr>
<th>Species</th>
<th>Location</th>
<th>Cu</th>
<th>Ag</th>
<th>Pb</th>
<th>Zn</th>
<th>Ni</th>
<th>Co</th>
<th>Mn</th>
<th>Fe</th>
<th>As</th>
<th>Cd</th>
<th>Cr</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td><em>Egeria radiata</em></td>
<td>Nigeria</td>
<td>---</td>
<td>---</td>
<td>0.3-3.6</td>
<td>96-172</td>
<td>0.6-4.5</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>0.1-0.6</td>
<td>---</td>
<td>---</td>
<td>Etim et al., 1991</td>
</tr>
<tr>
<td><em>Corbicula fluminea</em></td>
<td>Argentina</td>
<td>28-89</td>
<td>---</td>
<td>---</td>
<td>118-316</td>
<td>1-6</td>
<td>---</td>
<td>15-81</td>
<td>0.5-2.0</td>
<td>0.5-1.9</td>
<td>1.3-11</td>
<td>---</td>
<td>Bilos et al., 1998</td>
</tr>
<tr>
<td><em>Circentia callypsa</em></td>
<td>Qatar</td>
<td>8.35</td>
<td>3.0</td>
<td>1.4</td>
<td>69.1</td>
<td>23.9</td>
<td>4.4</td>
<td>17.7</td>
<td>517</td>
<td>156</td>
<td>1.1</td>
<td>0.9</td>
<td>de Mora et al., 2004</td>
</tr>
<tr>
<td><em>Saccostrea cucullata</em></td>
<td>Oman</td>
<td>60.9-276</td>
<td>---</td>
<td>0.3-0.6</td>
<td>391-1610</td>
<td>0.8-3.1</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>8.9-21.9</td>
<td>---</td>
<td>de Mora et al., 2004</td>
<td></td>
</tr>
<tr>
<td><em>Saccostrea cucullata</em></td>
<td>UAE, Oman Gulf</td>
<td>63.8</td>
<td>0.2</td>
<td>425</td>
<td>118-316</td>
<td>1-6</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>6.15</td>
<td>---</td>
<td>de Mora et al., 2004</td>
<td></td>
</tr>
<tr>
<td><em>Pinctada radiata</em></td>
<td>UAE, Persian Gulf</td>
<td>4.6-17.3</td>
<td>---</td>
<td>0.1-2.2</td>
<td>159-1430</td>
<td>0.5-7.0</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>2.7-9.9</td>
<td>---</td>
<td>de Mora et al., 2004</td>
<td></td>
</tr>
<tr>
<td><em>Pinctada radiata</em></td>
<td>Bahrain</td>
<td>3.1-4.4</td>
<td>---</td>
<td>0.3-3.9</td>
<td>1825-4290</td>
<td>0.7-0.8</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>3.6-3.7</td>
<td>---</td>
<td>de Mora et al., 2004</td>
<td></td>
</tr>
<tr>
<td><em>Perna perna</em></td>
<td>Senegal</td>
<td>7.2-7.7</td>
<td>---</td>
<td>---</td>
<td>64.5-121.6</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>0.7-2.3</td>
<td>---</td>
<td>---</td>
<td>Sidoumou et al., 2006</td>
<td></td>
</tr>
<tr>
<td><em>Crassostrea gasar</em></td>
<td>Senegal</td>
<td>47.1</td>
<td>---</td>
<td>2320</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>---</td>
<td>6.8</td>
<td>---</td>
<td>---</td>
<td>Sidoumou et al., 2006</td>
</tr>
<tr>
<td><em>Ruditapes decussus</em></td>
<td>Egypt*</td>
<td>23.8</td>
<td>---</td>
<td>15.5</td>
<td>155.6</td>
<td>21.5</td>
<td>23.5</td>
<td>24</td>
<td>961.5</td>
<td>---</td>
<td>4.5</td>
<td>---</td>
<td>Gabr &amp; Gab-Alla, 2008</td>
</tr>
<tr>
<td><em>Venrupis pallastra</em></td>
<td>Egypt*</td>
<td>23.2</td>
<td>---</td>
<td>18.5</td>
<td>132.2</td>
<td>27.5</td>
<td>30.5</td>
<td>26.5</td>
<td>1896.5</td>
<td>---</td>
<td>8.7</td>
<td>---</td>
<td>Gabr &amp; Gab-Alla, 2008</td>
</tr>
<tr>
<td><em>Comptopallium radula</em></td>
<td>New Caledonia</td>
<td>3.7-3.9</td>
<td>0.0-5.2</td>
<td>---</td>
<td>176-183</td>
<td>12.8-13.8</td>
<td>1.7-1.9</td>
<td>5.9-6.0</td>
<td>221-289</td>
<td>44.7-86.9</td>
<td>1.1.3.5</td>
<td>3.2</td>
<td>Metian et al., 2008</td>
</tr>
<tr>
<td><em>Isognomon isolognomon</em></td>
<td>New Caledonia</td>
<td>3.1-17.3</td>
<td>1.4-32.8</td>
<td>---</td>
<td>1718-13817</td>
<td>2.2-16</td>
<td>0.6-1.6</td>
<td>20.4-34.7</td>
<td>---</td>
<td>21.6-76.6</td>
<td>1.1-1.8</td>
<td>2.7-9</td>
<td>Hédouin et al., 2009</td>
</tr>
<tr>
<td><em>Gafrarium tumidum</em></td>
<td>New Caledonia</td>
<td>5.6-77.3</td>
<td>0.0-33.1</td>
<td>---</td>
<td>55.6-154</td>
<td>8.1-30.2</td>
<td>1.1-3.8</td>
<td>5.5-139</td>
<td>---</td>
<td>37.4-441</td>
<td>0.1-0.7</td>
<td>3.2-8</td>
<td>Hédouin et al., 2009</td>
</tr>
<tr>
<td><em>Solen dactylus</em></td>
<td>Iran, Northern Persian Gulf</td>
<td>13-20</td>
<td>0.3-1</td>
<td>4-7</td>
<td>60-92</td>
<td>2-8</td>
<td>5-16</td>
<td>8-56</td>
<td>210-1886</td>
<td>20-27</td>
<td>1-3</td>
<td>1-6</td>
<td>Present study</td>
</tr>
</tbody>
</table>

a: Contaminated Site
Table 5: Metal concentrations in marine sediment (µgg⁻¹ dry wt) and seawater (µgl⁻¹) in different locations of the Persian Gulf and guidelines

<table>
<thead>
<tr>
<th>Location/Standards</th>
<th>Cd</th>
<th>Pb</th>
<th>Ni</th>
<th>References</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sediment</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Global baseline values for metals in sediments</td>
<td>0.30</td>
<td>19</td>
<td>52</td>
<td>Bowen, 1979</td>
</tr>
<tr>
<td>RSA, northeastern Iranian coast</td>
<td>1.25</td>
<td>25</td>
<td>103</td>
<td>ROPME, 1999</td>
</tr>
<tr>
<td>Persian Gulf, Bahrain</td>
<td>0.40</td>
<td>12.30</td>
<td>15</td>
<td>ROPME, 1999</td>
</tr>
<tr>
<td>Persian Gulf and Oman Gulf</td>
<td>0.02-0.21</td>
<td>0.25-99</td>
<td>0.74-1010</td>
<td>de Mora et al., 2004</td>
</tr>
<tr>
<td>northwest Persian Gulf (Mahshahr Creeks)</td>
<td>0.27-1.00</td>
<td>7.09-29.72</td>
<td>65.57-171.41</td>
<td>Dehghan Madiseh et al., 2008</td>
</tr>
<tr>
<td>northern Persian Gulf</td>
<td>2.89</td>
<td>90.47</td>
<td>64.89</td>
<td>Pourang et al., 2005</td>
</tr>
<tr>
<td>northern Persian Gulf (Bandar Abbas)</td>
<td>&lt;0.50</td>
<td>4-7</td>
<td>29-46</td>
<td>Present study</td>
</tr>
<tr>
<td>Water</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>MPL for aquatic lifeᵃ</td>
<td>5</td>
<td>22</td>
<td>---</td>
<td>Gardiner &amp; Mance, 1984</td>
</tr>
<tr>
<td>ANZECC Guidelinesᵇ</td>
<td>2</td>
<td>5</td>
<td>15</td>
<td>Pourang et al., 2005</td>
</tr>
<tr>
<td>northern Persian Gulf</td>
<td>0.44</td>
<td>5.38</td>
<td>2.79</td>
<td>Pourang et al., 2005</td>
</tr>
<tr>
<td>northern Persian Gulf</td>
<td>0.83</td>
<td>1.11</td>
<td>8.43</td>
<td>present study</td>
</tr>
</tbody>
</table>

a: Maximum Permissible Levels
b: Australian Water Quality Guidelines for Fresh and Marine Waters, Australian and New Zealand Environment and Conservation Council (ANZECC)

**Discussion**

*Metal concentrations in biota, sediments and sea waters* The results showed that at two stations (Park-e-Qadir and Nakhel-e-Nakhoda) in Bandar Abbas (mean water temperature: 28.6±0.7°C; annual tidal range: -0.1- 3.88 m) the metal concentrations in razor clams, varied as follows:

Station 1:

\[
\text{Mg>Fe>Zn>Mn=As>Cu>Co=Ba>Pb>Ni>Cd>Cr>Ag}, \text{ and}
\]

Station 2:

\[
\text{Mg>Fe>Zn>Mn>As=Co=Cu>Ba>Pb=Ni>Cr>Cd>Ag}
\]

The trends are reasonably consistent with the literature values reported earlier (e.g., Etim et al. (1991), Zn>Ni>Pb>Cd for clam *E. radiata*, and Metian et al. (2008), Zn>Cd>Co=Mn for scallop *C. radula*). Taking into account that sediments are a sink for marine contaminants (Salomons et al., 1987) and their elements’ concentrations are often used to assess and monitor the contamination status. It is known that sediment-associated concentrations are not necessarily representative of the contaminant fraction (Hédouin et al., 2009). Therefore, to assess the difference in the contamination status of both stations, metal concentrations were analysed in clams, sediments and water (Tables 1 and 2). The concentrations of Mg and Fe in all clams and sediments and Mg and Cu in water at both stations were
more than other metals which showed that these metals were the most abundant metals in this area as their natural abundance in the earth’s crust. It is suggested that clams accumulate these metal elements in their body for some physiological requirements while bioaccumulating some of them more than their requirements according to their environmental pollutions (Bilos et al., 1998). This study indicated that this clam could be a valuable bioindicator for Zn after complementary and further studies; whereas, Metian et al. (2008) concluded that scallop Comptopallium radula would be a suitable bioindicator for Ag, As, Cd and Fe. S. dactylus is an edible clam (Saeedi et al. 2009); therefore, a study of the metal concentrations in this valuable species was necessary and vital for biomonitoring plans and management programs (as a bioindicator) and the risk assessment for humans (as an edible clam). According to low concentration of some metals in this clam comparing to its environment, this clam can be introduced as approximately safer seafood regarding other mentioned clams in other presented literatures. Clams accumulate Zinc in their bodies (54-107 µg g⁻¹ dry wt) more than sediments (14-22 µg g⁻¹ dry wt) and water (13-22 µg l⁻¹). Similar results were observed in the literature (de Mora et al. (2004) and Bilos et al., (1998)) for relatively high level of Cu and Zn accumulations in marine tissues while they did not follow the high concentrations of these metal elements in the environmental media. Cu and Zn metals are essential elements used in the structures of many metalloenzymes and metalloproteins such as haemocyanin and zinc fingers (Lippard and Berg, 1994). Based on the present findings and other studies, sessile bivalves such as S. dactylus, S. cucullata and C. fluminea distribute Zn metal in their bodies more than their environments. These metal distribution differences could be due to some physiological requirements, metabolic activities and body composition processes. However, the kind of phytoplankton which clams feed on is important for assimilation metals. As an illustration, in green mussel Perna viridis and the Manila clam Ruditapes philippinarum assimilated metals at a higher efficiency from the diatom diet (Thalassiosira pseudonana) than from inorganic sediment particles (Chong and Wang, 2000).

Mn, Pb, Ni and Fe concentrations in sediments were higher than in clams (Tables 1 and 2). De Mora et al. (2004) reported that S. cucullata concentrate lower values of Ni (0.80-3.14 µg g⁻¹ dry wt) than in the sediments (0.74-1010 µg g⁻¹ dry wt) in Oman and UAE. He suggests that nickel could not accumulate in those clams bodies; therefore, it does not play an integral role in their biological activities. In this study significant positive correlation were observed between the metal concentration in clams with sediments and water. S. dactylus is a filter feeder living in upper layers of sediments in a depth of 5-30 cm inside the canals (Saeedi et al., 2009). Therefore, some metals uptake from sediments and water into its body and accumulate in its tissues. To compare the result of the present work with other investigations, the concentrations of accumulated metals in S.
areas such as the Oman Sea were shown to contain elevated Cd concentrations (8.9-21.9 \( \mu gg^{-1} \) dry wt) (de Mora et al., 2004), such a bioaccumulation ability, presumably, does not appear to relate to anthropogenic contamination of the environment. Cadmium is a non-biodegradable and non-beneficial heavy metal whose role in the cell is not known yet (Lal Shah, 2010). As it is shown in table 4, both clams \textit{S. dactylus} (in this study) and \textit{E. radiata} (from the contaminated site of Nigeria) encompass the maximum values of Fe concentrations (1886 and 1896 \( \mu gg^{-1} \) dry wt) in their body respectively. However, Ag concentrations in \textit{S. dactylus} were the lowest among other clams reported in the literature (0.3-1 \( \mu gg^{-1} \) dry wt) (table 4). Ag concentrations were in studied sediment samples (~0.3 \( \mu gg^{-1} \) dry wt) which were lower than those reported earlier. Clams and sediments at station 2 contained more concentrations of Fe and Mn than station 1 which can show that concentration of metals in sediments and clams are correlated. Moreover, Fe and Mn showed a positive significant correlation in the clam body, whereas Etim et al. (1991) found no significant linear average relationship between any two of the metals studied. It can suggest that relationship between two metals in clam’s body can describe some biological interactions between two metals in the clam’s life.

According to table 5 the Cadmium concentration in sediment of northern Persian Gulf (all northern parts of the Persian Gulf) was 2.89 \( \mu gg^{-1} \) dry wt. (Pourang et al. 2005) which was higher than that in the Regional Organization for the Protection of the Marine Environment (ROPME) Sea Area guideline (RSA) and other locations of the Persian Gulf. In this study Cd, Pb and Ni concentrations (~0.50, 4-7 and 29-46 \( \mu gg^{-1} \) dry wt) (Bandar Abbas, Northern Persian Gulf) reached less values among RSA and other locations of the Persian Gulf, except Ni concentrations in Bahrain (15 \( \mu gg^{-1} \) dry wt). Pourang et al. (2005) demonstrated that the highest concentrations of Ni (64.89 \( \mu gg^{-1} \) dry wt) in sediments were observed close to the southern coast of Qeshm Island and Bandar Lengeh of northern parts of the Persian Gulf, Iran. The concentration of Ni in sediments of Bandar Abbas in this study (29-46 \( \mu gg^{-1} \) dry wt) was less than Qeshm and Bandar Lengeh which are close to Bandar Abbas. Also the concentration of these three metals in seawater showed the lower value among other studies except Cd and Ni in the study of Pourang et al. (2005) (Table 5).

Metal concentration in different clam body sizes

In this work, the effect of body size (age) on metal concentrations in the clam \textit{S. dactylus} has also been investigated. It is well known that body size reveals the metal concentrations in organisms and is related to the time courses of metal uptake. It is one of the most important parameters that reveal metal bioaccumulation in organisms (Hédouin et al., 2006 and 2009). The mechanisms of the effects of size are not fully understood; however, the body size is related to the age of the animal and more age means more time for
exposure to contaminants. But two

distinguished explanations are highlighted

below: (1) the decreasing surface/volume

ratio of an organism with increasing body

size, and (2) decreasing metabolic activity

in larger (older) organisms (Hédouin et al.,

2006). Both reasons would result in

decreasing metal uptake with increasing

individual size.

There was a positive significant
correlation between the clam’s length and
the concentrations of Cu and Mg in clams,
suggesting that the larger clams had a
higher ability to concentrate the metal than
smaller ones for Mg and Cu. Indeed, there
is a positive correlation between the age of
clams and the concentration of Cu and Mg.
It is more likely to be an increase in
physiological and biological requirements
in clams during their growth; however,
pollution levels and metal exposure time in
clams can affect the bioaccumulation
of some metals. Also according to the
previous results in this study Fe and Mg
had high concentrations in clams which
can suggest that these are two important
metals for clams’ life or because of
disability of clams to get rid of some
metals easily. A significant negative
relationship between size of clam
Gafrarium tumidum in New Caledonia and
concentrations of Cd, Cr, Co and Zn was
reported by Hédouin et al. (2006) whereas
the accumulation of Ag was positively
 correlated with size of clams. Bilos et al.
(1998) demonstrated that Cu levels in
Asiatic clam C. fluminea of Argentina
significantly correlated with their size
(positive) which indicated variable
physiological requirements for this metal
with age and showed a similar result of
this study. Additionally, in view of Bilos
et al. (1998) literature on Ag in bivalves,
the latter observations suggest the
occurrence of a specific detoxification
mechanism that would be more efficient in
older clams.

Metal concentration in different clam
sexes

Male and female clams were studied
separately for their metal concentrations.
There was no significant difference
between males and females. This species is
a gonochoric clam which spends the
ripeness stage of gametogenesise cycle in
January and February and in this stage
gonads are full of gametes whereas in
September and October clams are in the
first stages of gametogenesis with
approximately empty gonads (Saeedi et al.,
2009). When clams were analyzed for
some metals concentration in two different
sexual stages, a significant difference were
observed after ( stages I and II) and before
spawning (stage III A). It would suggest
that increasing the weight of clams and
gametogenic materials can affect the metal
concentrations in clams.

Metal concentration in different stations
and seasons

Among the two factors considered
(seasons and sampling sites) and their
interactions, both factors determined the
variability observed for all metals.
Significant interaction between two
seasons and two sampling stations was
detected for all metals, suggesting that
anthropogenic interference differences of
measured concentrations were dependent
upon the seasonal variation. Furthermore,
distribution and variation of some metals
in different stations is related to
environmental factors and anthropogenic activities.

Only Cu, Fe and Ba sediment concentrations showed significant differences in the first and second station, surprisingly with maximum concentrations in the second station. Solely Fe concentrations in seawater presented a significant difference between two stations, with a maximum value at station 2. Metal accumulations were higher in station 2 (620-820 μg l⁻¹). It might be related to some natural factors like fewer water currents and different coastal topography. Also we need to attach so much importance to a bus terminal near station 2, while station one is only next to a park with litters from people. All these factors can make station 2 more polluted for some metals than station 1.

Even if these results are somewhat surprising, they demonstrate that the two sampling sites are consistent with the metal concentrations in the clams which can suggest clam *S. dactylus* as a reliable bioindicator for this area after several further studies on other organisms and the ecosystem.
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References


بررسی میزان تجمع برخی فلزات سنگین در دوکفه ای، رسوبات و آب در ساحل گلشهر، بندرعباس، خلیج فارس

S. dactylus (von Cosel, 1989)

هامیه سعیدی ۱، آریا اشجع اردلان ۲، بهرام حسن‌زاده کیابی ۳، رامین زیبا سرست ۴

چکیده

در این مطالعه برشی فلزات شامل Cu، Ag، Pb، Zn، Ni، Co، Mn، Fe، As، Cd، Cr، Mg، Ba در دوکفه ای S. dactylus و رسوبات و آب در ساحل گلشهر، بندرعباس، خلیج فارس از ماده مورد بررسی قرار گرفت. نمونه برداری از دو ایستگاه (ایستگاه اول، پارک غدیر، / ۲۰ ۵۶ طول شرقی و ۲۰ ۰۷ عرض شمالی و ایستگاه دوم، نخل ناهدا، / ۲۲ ۳۶ طول شرقی و ۲۰ ۰۷ عرض شمالی) در ماه‌های آبان و بهمن ۱۳۸۸ انجام یافت. با توجه به این که مطالعه محدوده بر روی بررسی تجمع فلزات در مکان‌های وجود دارد، تحقیق حاضر اولین گزارش از تجمع فلزات سنگین در این دوکفه ای و Fe و Mg در این فلزات تجمع یافت. معرفي آن به عنوان یک شاخه زیستی است. نتایج این مطالعه نشان داد که در دو فلزی Fe و Mg در سطح بهتری هستند در حالی که در دو فلزی Ag و Zn در دوکفه ای بیشتر از در سطح بهتری هستند. این حال، این مطالعه حاضر اولین گزارش از تجمع فلزات در این دوکفه ای می‌تواند به عنوان یک شاخه زیستی مناسب برای فلز روی (Zn) معرفی گردد.

واژگان کلیدی: فلز، تجمع زیستی، شاخه زیستی، S. dactylus، رسوب، آب، خلیج فارس

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